

**Determination of the Chemical
Oxygen Demand COD according
to DIN 38 409, part 41**

Application

Use

Determination of the chemical oxygen demand above concentration of 15 mg/l. In the case of undiluted samples the procedure can be used up to 300 mg/l. The chloride contents of the undiluted sample must not exceed 1 g/l. Apart from that, the work should be performed according to the procedure laid down in Part 41-2.

Appliances

- Titrator: TL 6000/7000 (TL 6000/7000 M1/20 or TL 6000/7000 M1/50) consists of
- Basic device
- Magnetic stirrer TM 235
- 20 mL exchange unit WA 20 or 50 mL exchange unit WA 50, with amber glass bottle for the titrant, complete

Electrodes

- Electrode: Pt 61
- Titration tip TZ 1623
- Option: longer stand rod with (450 mm) Z 601

Reagents

- Titrant: Ferrous (II) ammonium sulphate solution 0.120 mol/l
- Sulphuric acid, potassium hydrogen phthalate solution as standard, potassium dichromate solution (containing mercuric sulphate) 0.02 mol/l

Description

Preparation of the titration solution

47.1 g of ferrous (II) ammonium sulphate are dissolved in water in a 1 l volumetric flask. 20 ml of sulphuric acid = 1.84g/l are added, solution is topped up with water to 1000 ml.

Titer standardisation of the titration solution

The titer of the solution is determined in use on a daily basis. To do so, 10.0 ml of the potassium dichromate solution 0.02 mol/l are diluted with water to approx. 100 ml. 30 ml of sulphuric acid are added carefully to this solution. After cooling down the solution is placed on the magnetic stirrer and titrated using 3 times the method „Titer COD“. A triple determination is carried out. The average value is automatically calculated in mol/l and stored into the exchange unit.

Application

Method Titer COD:

Method data

Method name:	COD Titer	Created at:	05/16/13 16:18:56
Method type:	Automatic titration	Last modification:	05/16/13 16:22:11
Measured value:	mV		
Titration mode:	Linear	Documentation:	GLP
Linear steps:	0.020 ml		

Measuring speed / drift:	User-defined:	minimum holding time:	01 s
		maximum holding time:	10 s
		Measuring time:	01 s
		Drift:	30 mV/min

Initial waiting time:	0 s		
Titration direction:	Decrease		
Pretitration:	9.000 ml	Delay time:	25 s
End value:	Off		
EQ:	On		
Slope value:	User-defined	Value:	1200

Dosing parameter

Dosing speed:	100.00 %	Filling speed:	30 s
Maximum dosing volume:	20.00 ml		

Calculation formula

Titer:	$(V * F2) / ((EQ1 - B) * M * F1) \rightarrow WA$	Mol (M):	1.00000
Unit:	mol/l	Decimal places:	4

Pattern (V):	1.000 ml	Factor 2 (F2):	1.2000
Blank value (B):	0.0000 ml	Factor 1 (F1):	1.0000
Statistics:	3		

Application

Titration

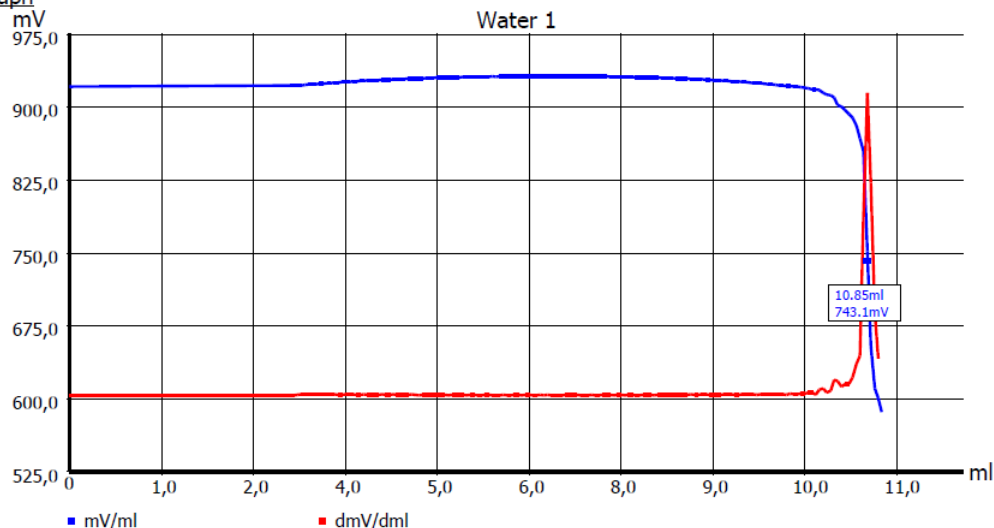
After the proper addition of potassium dichromate solution (containing mercuric sulphate) and sulphuric acid to the samples and the blank values, they are placed in a **suitable device** (e.g. from the company Behr) where the reaction mixture is heated up and boiled for 2 hours at 120 °C. Afterwards the samples and blank values are brought to room temperature and then placed on the magnetic stirrer. Carry out the method „COD blank“ first. Repeat the method 2 times. The three blank values are used to automatically calculate an average value which is stored into the global memory (M01: COD blank). Now the sample titrations can be carried out.

A control determination can be made using potassium hydrogen phthalate solution. In this case one uses 20 ml of the potassium hydrogen phthalate solution instead of the sample as pattern. The theoretical COD value of this solution is 200 mg/l. Values between 192 and 208 mg/l are admissible.

Result example

GLP documentation

Titration graph



Method data

Method name:	COD		
End date:	31.10.11		
End time:	16:28:46	Titration duration:	8 m 22 s

Titration data

Sample ID:	Water 1	Pattern:	20.0000ml
Start mV:	919.7mV	End mV:	586.4mV

EQ1: 10.854 ml/743.1mV

Application

Method

COD Blank value:

Method data

Method name:	COD Blank	Created at:	05/16/13 16:23:24
Method type:	Automatic titration	Last modification:	05/16/13 16:23:57
Measured value:	mV		
Titration mode:	Linear	Documentation:	GLP
Linear steps:	0.020 ml		

Measuring speed / drift:	User-defined:	minimum holding time:	01 s
		maximum holding time:	10 s
		Measuring time:	01 s
		Drift:	30 mV/min
Initial waiting time:	0 s		
Titration direction:	Decrease		
Pretitration:	9.000 ml	Delay time:	25 s
End value:	Off		
EQ:	On		
Slope value:	User-defined	Value:	1200

Dosing parameter

Dosing speed:	100.00 %	Filling speed:	30 s
Maximum dosing volume:	20.00 ml		

Calculation formula

Blank:	EQ1 -> M01		
Unit:	ml	Decimal places:	2

Statistics:	3
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Application

COD sample:

Method data

Method name:	COD sample	Created at:	05/16/13 16:25:15
Method type:	Automatic titration	Last modification:	05/16/13 16:26:42
Measured value:	mV		
Titration mode:	Linear	Documentation:	GLP
Linear steps:	0.040 ml		

Measuring speed / drift:	User-defined:	minimum holding time:	01 s
		maximum holding time:	10 s
		Measuring time:	01 s
		Drift:	30 mV/min
Initial waiting time:	0 s		
Titration direction:	Decrease		
Pretitration:	3.000 ml	Delay time:	25 s
End value:	Off		
EQ:	On		
Slope value:	User-defined	Value:	1200

Dosing parameter

Dosing speed:	100.00 %	Filling speed:	30 s
Maximum dosing volume:	20.00 ml		

Calculation formula

COD:	$(B-EQ1)*T*M*F1/(V*F2) \rightarrow M01$	Mol (M):	1.00000
Unit:	mg/l	Decimal places:	1

Blank value (B):	M01	Titre (T):	auto
Factor 1 (F1):	8000.0000	Pattern (V):	Vol
Factor 2 (F2):	1.0000	Statistics:	3

Application

Electrode handling

After use, the Pt 61 electrode should be placed in KCl 3 mol/l immediately. It has to be made sure that there is always sufficient electrolyte solution in the electrode. For further details, please refer to the electrode's operating instructions.

Notes

If you have any questions on the application, you can feel free to contact us..

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